

CHEM 531: ADVANCED INORGANIC CHEMISTRY
SECTION 01W---CRN#21185
COURSE SYLLABUS: SPRING 2014

Instructor: **Dr. Bukuo Ni**

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Office Hours: Wednesday: 1:30 – 2:30 pm, Friday: 11:00am – 12:00pm or by appointment.

COURSE INFORMATION

Lectures (Web Based Class): 1/30/14 through 5/09/14

Required Textbook: "Inorganic Chemistry", Catherine E. Housecroft and Alan G. Sharpe (3rd or 4th Edition, ISBN 978-0-13-175553-6 or 978-0-273-74275-3)

Additional Book for your study of this course:

"Inorganic Chemistry", 5th Edition, Shriver & Atkins, ISBN: 1-4292-1820-7.

Course description: This course covers descriptive chemistry of more interesting elements and compounds and the standard topics in coordination, organometallic, solid-state chemistry, and catalysis and some industrial processes. Prerequisites are sound knowledge of general chemistry and familiarity with organic, physical and analytical chemistry. Regular attendance and active learning are expected. Students' questions and comments are welcome.

Student learning outcomes: At the end of the course, the student will be able; (1) to describe and explain the coordination compounds containing metal as central atom which surrounded by ligands; (2) to understand the stereochemistry of coordination compounds; (3) to classify the type and mechanism involve in coordination compounds reactions; (4) to study the characterization of coordination compounds and its application.

ACCESS AND NAVIGATION

This course is an eCollege online supporting course (a web based course). eCollege is a campus wide web-enhanced internet teaching and learning support system. Students taking this course will be able to surf online course website, get reading material, download and upload assignments, check grades and cumulative points with percentiles anytime online.

If you are not familiar with the use of eCollege or the Library Online Services, please avail yourself of the online tutorials which is available through your MyLeo web page.

You will need your CWID and password to log in to the course. If you do not know your CWID or have forgotten your password, contact Technology Services at 903.468.6000 or helpdesk@tamuc.edu

GRADING

Your performance and final grade in the course will be evaluated on the basis of total points earned. The distribution of points will be based on the following: Homework (30 points), which will be assigned and discussion throughout the semester. Midterm exam and final exam will carry 35 points, respectively, with total 70 points. The exams will be taken on Rockwall and Commerce campus,

respectively. The final letter grade will be based on a standard scale 90-100% A, 80-89% B, 70-79% C, 60-69% D, and below 60% F. The grades may be curved, if warranted.

There will be absolutely no make-ups for exams except extraordinary circumstance (notes required). If you miss an examination, you will be assigned a zero for that assignment. Homework not submitted on time may receive a grade of zero.

Academic Integrity Code:

Ethical behavior is expected in all work. Any material submitted in Inorganic Chemistry must represent your own work and follow the Academic Integrity Code. Students supplying materials for others to "look at" (e.g. exams) may be charged with academic misconduct. The use of 'cheat sheets', stored text, constants, or formulas in calculators may be regarded as a violation of academic standards. A zero tolerance policy will be in effect. If you haven't already done so, you should familiarize yourself with TAMU-C's academic policies and regulations, especially those dealing with academic integrity.

Tentative Schedule

The tentative schedule is subject to change.

<i>Week of</i>	<i>Lecture Topic</i>	<i>Reading</i>
1 Jan 30-Feb. 7	An introduction to molecular symmetry	Ch. 4
2 Feb 8-16	Acids, bases and ions in aqueous solution	Ch. 7
3 Feb 17- 23	Groups 1& 2: the alkali & alkaline earth metals	Ch. 11&12
4 Feb 24-Mar. 2	The group 13 elements	Ch. 13
5 Mar. 3-9	The group 14 &15 elements	Ch. 14&15
SB Mar. 10-16	Spring break	
6 Mar. 17-23	The group 16 elements (Midterm exam: Ch. 4-15)	Ch. 16
7 Mar. 24-30	The group 17 elements	Ch. 17
8 Mar. 31-Apr. 6	d-Block metal chemistry: general consideration	Ch. 20
9 Apr. 7-13	d-Block metal chemistry: coordination complex	Ch. 21
10 Apr. 14-20	Organometallic compounds of d-block elements	Ch. 24
11 Apr. 21-27	d-Block metal complexes: reaction mechanisms	Ch. 26
12 Apr. 28-May 4	Catalysis and some industrial processes	Ch. 27
13 May 5-9	Final exam: Ch. 16-27	

ADA ELIGIBLE STUDENTS: Students requesting accommodations for disabilities must make arrangements through the Disability Resources & Services office. For more information, please contact the Office of Student Disability Resources and Services, Texas A&M University-Commerce, Gee Library Room 13, Phone (903) 886-5150 or (903) 886-583, Fax (903) 468-8148, E-mail: StudentDisabilityServices@tamuc.edu. ADA eligible students should make arrangements with the instructor in the first week of the semester regarding special arrangements needed for classroom or testing facilities and procedures to accommodate the disability.

* Please note that this schedule and topics are subject to change